



# High removal of crystal violet dye and tetracycline by hydrochloric acid assisted hydrothermal carbonization of sugarcane bagasse prepared at high yield

Farahin Mohd Jais<sup>a</sup>, Shaliza Ibrahim<sup>b,\*</sup>, Ching Yern Chee<sup>c</sup>, Zubaidah Ismail<sup>a</sup>

<sup>a</sup> Department of Civil Engineering, Faculty of Engineering, University of Malaya, Kuala Lumpur, 50603, Malaysia

<sup>b</sup> Institute of Ocean and Earth Science, University of Malaya, Kuala Lumpur, 50603, Malaysia

<sup>c</sup> Department of Chemical Engineering, Faculty of Engineering, University of Malaya, Kuala Lumpur, 50603, Malaysia

## ARTICLE INFO

### Keywords:

Adsorption  
Antibiotic removal  
Dye removal  
Hydrochar  
Hydrothermal carbonization  
Response surface methodology

## ABSTRACT

At a moderate thermal treatment process, hydrothermal carbonization (HTC) was known as an alternate and green way of preparing carbonaceous material known as hydrochar as an adsorbent. The HTC process requires the inclusion of water as a carbonization medium for the hydrolysis reaction to occur. By adding acid to the HTC water, the hydrolysis reaction was catalyzed, which lowered the reaction time and temperature while also increasing the adsorption efficiency. Overreaction, on the other hand, may occur, lowering the yield as well as the number of functional groups accessible for adsorption. Thus, in this study, Response Surface Methodology by Central Composite Design (RSM-CCD) was employed to investigate and optimize the HCL acid-assisted HTC of sugarcane bagasse (SB) process parameters of loading rate, reaction time, reaction temperature, and HCL acid concentration towards three responses: optimized hydrochar (HC<sub>op</sub>) yield, crystal violet (CV) dye removal and tetracycline (TC) removal. The HC<sub>op</sub> was characterized using FTIR, FESEM, BET + N<sub>2</sub> gas, and TGA analyses, and the results were compared to hydrochar made without the use of acid (HC<sub>dw</sub>). In brief, HC<sub>op</sub> has more acidic oxygenated functional groups, stronger aromaticity and hydrophobicity, greater porosity, and greater thermal stability than HC<sub>dw</sub>. The Langmuir isotherm model reported maximum adsorption capacity ( $Q_{max}$ ) of CV dye removal for HC<sub>op</sub> (207.16 mg g<sup>-1</sup>) was 1.5 times higher than the HC<sub>dw</sub> (137.85 mg g<sup>-1</sup>). While HC<sub>op</sub> had a two-fold higher  $Q_{max}$  of TC removal (68.25 mg g<sup>-1</sup>) than HC<sub>dw</sub> (33.61 mg g<sup>-1</sup>). Overall, the optimization of HCL-acid assisted HTC of SB was successful in producing hydrochar with good adsorption efficiency.

## List of symbols & abbreviations

Å angstrom  
A<sub>HJ</sub> Harkins-Jura isotherm constant  
ANOVA analysis of variance  
A<sub>T</sub> maximum binding energy

\* Corresponding author.

E-mail address: [shaliza@um.edu.my](mailto:shaliza@um.edu.my) (S. Ibrahim).