



## Microstructural characterization and reaction mechanism of home-brewed activator derived from eco-processed pozzolan for one-part geopolymer mortar

Gokulanathan Venkatesan<sup>a,✉</sup>, U. Johnson Alengaram<sup>a,\*</sup>, Shaliza Binti Ibrahim<sup>b,c</sup>,  
Muhammad Shazril Idris Bin Ibrahim<sup>a,b</sup>, Sharifah Binti Mohamad<sup>d</sup>

<sup>a</sup> Centre for Innovative Construction Technology (CICT), Department of Civil Engineering, Faculty of Engineering, Universiti Malaya, Kuala Lumpur 50603, Malaysia

<sup>b</sup> Institute of Ocean and Earth Science (IOES), Universiti Malaya, Kuala Lumpur 50603, Malaysia

<sup>c</sup> Renewable energy, Institute of Sustainable energy, Universiti Tenaga Nasional (UNITEN), Kajang, Selangor 43000, Malaysia

<sup>d</sup> Department of Chemistry, Faculty of Science, Universiti Malaya, Kuala Lumpur 50603, Malaysia

### ARTICLE INFO

#### Keywords:

Alkali-fusion method  
Eco-processed pozzolan  
Home-brewed activator  
Reaction mechanism  
carbon efficiency  
One-part geopolymers

### ABSTRACT

Anhydrous sodium-based commercial activators for one-part geopolymer (OGP) may cause eye damage, skin burns due to their corrosive nature, highly hygroscopic nature, high carbon footprint during production, high energy consumption, and cost. This research developed a home-brewed activator (HBA) from industrial/agro-based ash such as Eco-processed pozzolan (EPP) using an alkali fusion method. In the alkali fusion method, a mixture of sodium hydroxide (NaOH) powder and EPP at various weight ratios of 1, and 2 was calcinated in a muffle furnace at 300 °C and 500 °C for 1.5 and 3 h. Ground Granulated Blast Furnace Slag and Fly ash were used as binders for casting the OGP and two-part geopolymer (TGP) mortar specimens. From the X-ray diffraction (XRD) results, the main activator compounds such as thermonatrite ( $\text{Na}_2\text{CO}_3 \cdot \text{H}_2\text{O}$ ), natrite ( $\text{Na}_2\text{CO}_3$ ), calcite ( $\text{CaCO}_3$ ) and natrosilicate ( $\text{Na}_2\text{Si}_2\text{O}_5$ ) in HBA that initiate the polymerization reaction. The OGP mortar specimen produced a 28-day compressive strength of 50 MPa under the ambient curing regime with a NaOH/EPP ratio of 1, calcination temperature of 300 °C, and duration of 1.5 h. Both the OGP and TGP mortar specimens showed the presence of calcium aluminosilicate hydrate and calcium sodium aluminosilicate hydrate (C-A-S-H/C-N-A-S-H) geopolymeric gel. The carbon efficiency of OGP mortar specimens was found 70 % lower than that of TGP mortar specimens.

**Abbreviations:** ~, approximately equal; °C, degree Celsius; %, percentage;  $\mu\text{m}$ , micrometre; A<sub>c</sub>, Amorphous content; CC, Calcite; Cp, Specific heat capacity; C/S, Cross section; CS, Control specimens; D<sub>c</sub>, Degree of crystallinity; g, gram; J, Joule; K<sub>b</sub>, basicity coefficient; KN, kilo Newton; KV, kilo Volt; mA, milli Ampere; mg, milli gram; ml, milli litre; min, minutes; mPa.s, milli Pascal second; M, Molarity; M<sup>+</sup>, Metal ion; N, Natrite; NH, Sodium hydroxide; NS, Natrosilicate; Q, Quartz; T, Thermonatrite; T, tonnes; W, Watts; w/b, water/binder; AAMs, Alkali activated materials; ASTM, American Standard for Testing and Materials; C-A-S-H, Calcium aluminosilicate hydrate; COD, Crystallography open Database; C-N-A-S-H, Calcium sodium aluminosilicate hydrate; DTG, Differential thermo gravimetry; EPP, Eco-processed pozzolan; FA, Fly ash; FESEM, Field Emission Scanning Electron Microscopy; EDS, Energy Dispersive Spectroscopy; EDX, Energy Dispersive X-ray; FM, Fineness Modulus; FTIR, Fourier Transform Infrared; GGBFS, Ground Granulated Blast Furnace Slag; GP, Geopolymers; GPC, Geopolymer concrete; HBA, Home-brewed activator; Hours, h; LOI, Loss on ignition; MPa, Mega Pascal; N-A-S-H, Sodium Aluminosilicate Hydrate; NBO, Non-bridging oxygen; N-C-A-S-H, Sodium calcium aluminosilicate hydrate; OGP, One-part geopolymer; OPC, Ordinary Portland cement; POFA, Palm oil fuel ash; TGP, Two-part geopolymer; ppm, parts per million; RHA, Rice Husk Ash; SAF,  $\text{SiO}_2 + \text{Al}_2\text{O}_3 + \text{Fe}_2\text{O}_3$ ; SBA, Sugarcane Bagasse Ash; SBE, Spent bleaching earth; SBEA, Spent bleaching earth ash; SE, Secondary electron; SEM, Scanning Electron Microscopy; SF, Silica Fume; SM, Silicate Modulus; SS, Sodium Silicate; SSA, Specific Surface Area; TGA, Thermo-gravimetric analysis; TPOFA, Treated palm oil fuel ash; WD, Wavelength -dispersive; XRD, X-ray Diffraction; XRF, X-ray Fluorescence; Al, Aluminium;  $\text{Al}_2\text{O}_3$ , Aluminium oxide; Ca, Calcium; CaO, Calcium oxide;  $\text{CaCO}_3$ , Calcium carbonate;  $\text{CaSO}_4$ , Calcium sulfate;  $\text{CO}_2$ , Carbon dioxide; Fe, Iron;  $\text{Fe}_2\text{O}_3$ , Ferric oxide; KOH, Potassium hydroxide;  $\text{K}_2\text{O}$ , Potassium oxide; MgO, Magnesium oxide; NaOH, Sodiumhydroxide;  $\text{NaAlO}_2$ , Sodium aluminium oxide;  $\text{Na}_2\text{O}$ , Sodium oxide;  $\text{Na}_2\text{CO}_3$ , Sodium carbonate;  $\text{Na}_2\text{CO}_3 \cdot \text{H}_2\text{O}$ , hydrated sodium carbonate;  $\text{Na}_2\text{SiO}_3$ , Sodium silicate;  $\text{Na}_2\text{SiO}_3 \cdot \text{H}_2\text{O}$ , Hydrated sodium metasilicate;  $\text{Na}_2\text{Si}_2\text{O}_5$ , Natro silicate;  $\text{Na}_2\text{SO}_4$ , Sodium sulfate; Si, Silica;  $\text{SiO}_2$ , Silicon oxide;  $\text{SO}_3$ , Sulfur trioxide.

\* Corresponding author.

E-mail address: [johnson@um.edu.my](mailto:johnson@um.edu.my) (U.J. Alengaram).

<https://doi.org/10.1016/j.conbuildmat.2025.140513>

Received 7 December 2024; Received in revised form 18 February 2025; Accepted 19 February 2025

Available online 25 February 2025

0950-0618/© 2025 Elsevier Ltd. All rights are reserved, including those for text and data mining, AI training, and similar technologies.