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NiMoO₄ nanoparticles decorated carbon nanofiber membranes for the flexible and high performance glucose sensors



S. Divya Rani^a, R. Ramachandran^{a,*}, Sunirmal Sheet^b, Md. Abdul Aziz^c, Yang Soo Lee^b, Abdullah G. Al-Sehemi^d, Mehboobali Pannipara^d, Yang Xia^e, Shu-Yi Tsai^f, Fong-Lee Ng^g, Siew-Moi Phang^{g,h}, G. Gnana kumar^{i,*}

- ^a The Madura College, Department of Chemistry, Vidya Nagar, Madurai, 625 011, Tamil Nadu, India
- b Department of Forest Science and Technology, College of Agriculture and Life Science, Chonbuk National University, 567 Baekje-daero, Jeonju-si, 54896, Jeollabuk-do, Republic of Korea
- ^c Centre of Research Excellence in Nanotechnology (CENT), King Fahd University of Petroleum and Minerals (KFUPM), Dhahran, 31261, Saudi Arabia
- ^d Department of Chemistry, King Khalid University, Abha, 61413, Saudi Arabia
- e College of Materials Science and Engineering, Zhejiang University of Technology, Hangzhou, 310014, China
- Department of Materials Science and Engineering, Hierarchical Green-Energy Materials Research Center, National Cheng Kung University, Tainan, Taiwan
- g Institute of Ocean and Earth Sciences (IOES), University of Malaya, 50603, Kuala Lumpur, Malaysia
- ^h Faculty of Applied Sciences, UCSI University, Cheras, 56000, Kuala Lumpur, Malaysia
- ⁱ Department of Physical Chemistry, School of Chemistry, Madurai Kamaraj University, Madurai, 625021, Tamil Nadu, India

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ABSTRACT

The free-standing and flexible carbonaceous nanofiber membrane (CNF) comprising of NiMoO₄ nanoparticles decorated carbon fibers were developed for glucose sensing analysis. The applicability of polymeric nanofiber membranes in electrochemical sensors is accelerated by enhancing their electrical conductivity via carbonization process. The cavities exist in CNFs accelerate the rapid diffusion of glucose and maximizes the analyte utilization efficiency. The uniform implantation of catalytic active sites of NiMoO₄ on CNFs accelerates the glucose sensing kinetics further. With the synergism of bimetal active sites, porous architecture, and conductive carbon network, NiMoO₄/CNF demonstrates the excellent sensitivity, low detection limit, and broad linear range, respectively, of $301.77 \,\mu\text{A} \,\text{mM}^{-1} \,\text{cm}^{-2}$, $50 \,\text{nM}$, and 0.0003– $4.5 \,\text{mM}$ toward glucose sensing. Also, the sensing concerts of NiMoO₄/CNF are reliable, repeatable, and electrochemically stable along with the high specificity and real sample applicability toward glucose detection. Thus, the free-standing, bendable, binder-free, reusable, and cost-efficient fabrication process developed for the NiMoO₄/CNF membrane realizes it's conscription in the development of cost-efficient and high performance glucose sensors.

1. Introduction

Glucose is deemed as an essential nutrient to promoting the metabolism process and physiological activities of living organisms [1]. However, an improper glucose level in human body leads to the metabolic disorders of insulin secretion reduction (Diabetes Mellitus - Type 1 (DM1)) and insulin resistance (Diabetes Mellitus - Type 2 (DM2)) [2]. It causes major health afflictions including peripheral arterial, diabetes mellitus, kidney failure, obesity, and nerve degeneration [3]. Hence, the development of cost efficient, accurate, and reliable glucose sensors is emphasized as the prime of hour in various fields including clinical diagnostics [4], ecological [5], food [6], biotechnology [7], and pharmaceutical industries [7]. Amid the various techniques adopted for the

detection of glucose, enzymeless electrochemical technique is prevalently recommended owing to its simple handling, environmental friendliness, cost-effective platform, high sensitivity, and good selectivity [8,9].

Although various electrodes including glassy carbon electrode (GCE) [10], Indium titanium oxide (ITO) [11], Fluorine-doped titanium oxide (FTO) [12], carbon cloth [13], gold [14], and platinum [15] were widely used in glucose sensors, their disadvantages including conventional polishing, pre-treatment and modification processes, use of inactive binder, inferior stability, large potential, and lower sensitivity hinder their prospective applications [16]. It urges the development of free-standing electrodes and their direct modification with the catalytic nanostructures. Recently, our group has demonstrated the use of

E-mail addresses: ultraramji@gmail.com (R. Ramachandran), kumarg2006@gmail.com (G.G. kumar).

^{*} Corresponding authors.